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Toward development of optical biosensors based on photoluminescence of TiO₂ nanoparticles for the detection of *Salmonella*

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Abstract

Quality control of food and agriculture production is an inseparable part of human safety and well-being. *Salmonella* infections belong to one of the most monitored pathogens in the world, therefore advanced determination of this pathogen can decrease the risks of human diseases caused by this microorganism. In this research we introduce a novel optical immunosensor for determination of *Salmonella typhimurium*. The immunosensor is based on Titanium dioxide (TiO₂) nanoparticles deposited on glass substrates (glass/TiO₂). TiO₂ nanoparticles exhibit an intense photoluminescence (PL) in the visible range of spectrum at room temperature. The direct immobilization of antibodies (anti-S-Ab) against *Salmonella* antigens on the surface of glass/TiO₂ has resulted in the formation of glass/TiO₂/anti-S-Ab-based structure, which was characterized by increased PL intensity and IR-shifted position of the PL peak in comparison to the same characteristics of glass/TiO₂-based structure. The changes of the PL intensity and peak positions after interaction of the immobilized anti-S-Ab with *Salmonella* antigens (*Salmonella*-Ag) were used as immunosensor signal, allowing sensitive and selective detection of *Salmonella*-Ag in a label-free configuration. The sensitivity of the reported optical immunosensor towards *Salmonella*-Ag is in the range from 10³ to 10⁵ cell/ml. Some aspects of the interaction between TiO₂ and biological compounds have been discussed. This work opens up new possibilities for

the development of optical label-free immunosensors suitable for fast, simple and efficient analysis of *Salmonella* infections.

Keywords: TiO₂ nanoparticles, Photoluminescence based biosensor, Immunosensor, *Salmonella* infection, Antibody

Introduction

TiO₂ is a well-known material, which was used for the application in different sensors and biosensors . TiO₂ shows good stability in chemically aggressive environment what makes it an attractive material for sensors acting in such environments . TiO₂ is a wide band gap semiconductor with indirect optical transitions . Quantum confinement effects in nanostructured TiO₂ resulted in the increase of the band gap, the enhancement of photocatalytic activity and intense photoluminescence (PL) at room temperature .

A number of papers have reported the PL of TiO₂-based nanostructures at room temperature . It was found that some TiO₂-based nanostructures demonstrated emission in the range of 430-560 nm . Two main mechanisms of PL for TiO₂-based nanostructures were proposed: one is based on recombination of self trapped excitons (STE) (430-530 nm) and the other one is based on oxygen vacancies (530-560 nm) . The recent papers on PL of TiO₂ and the influence of dopants and type of nanostructure (nanotubes, nanosheets, etc.) point to a significant interest in this material and prospects of using TiO₂ PL in different sensors, especially in biosensors . TiO₂ has low isoelectric point pH=5.5 what is advantageous for protein immobilization on its surface . Due to this property the TiO₂ was used in adsorbed enzyme based electrochemical biosensors .

Optical detection methods such as absorbance, reflectance and photoluminescence are suitable for simple, fast and accurate detection of target analytes . In the PL spectra of TiO₂ were studied under different concentrations of NO₂ at room temperatures in order to detect NO₂ with TiO₂-based optical gas sensor. An increase of the PL intensity was registered and no shift of PL peak position was observed. It was reported that NO₂ adsorption on the sensor surface stimulated irradiative transitions in TiO₂ .

Quenching of the PL of TiO₂-based quantum dots by BSA and DNA has been performed showing that the concentration of these biomolecules has strong influence on the decrease of PL intensity . In this case also no peak shifts were observed. This research illustrated that the PL of TiO₂-based nanostructures could be applied for determination of various biomolecules (including

DNA and proteins). TiO₂ nanotubes were used for optical interferometry based label free sensing of rabbit immunoglobulin G (IgG) . In this case a protein A capture probe was used, which was immobilized on the inner walls of TiO₂ nanotubes by electrostatic adsorption. The selectivity of interaction was confirmed by experiments using IgG from poultry, which did not bind to protein A .

An immunosensor is a type of biosensors based on specific interaction between antibody (Ab) – antigen (Ag) immune complex . The Ab and Ag interaction is mostly very specific and sensitive towards some analytes what makes immunosensors suitable for accurate and precise tests with electrochemical, optical, magnetic and piezoelectric transducers . Some recent results show an efficient application of nanomaterials (e.g. nanoparticles and quantum dots) in biosensor platforms suitable for the detection of pathogens on the basis of surface plasmon resonance (SPR) and fluorescence based methods .

Food safety and agriculture are important fields of human activities . The quality control of food and agriculture production will decrease the risks of human diseases, caused by toxin infections. *Salmonella typhimurium* is one of the most monitored pathogens in EU . Some optical biosensors, based on SPR and reflectance spectroscopy, were developed for the detection of *Salmonella* at concentrations from 10² to 10⁵ cells/ml .

In our previous papers we have shown the application of photoluminescent nanostructures for the design of immunosensor . In the present work we report the application of TiO₂ nanoparticles in the design of PL-based immunosensor suitable for the detection of *Salmonella*.

Experimental

1. Preparation of TiO₂-NP-modified samples

Anatase nanoparticles with average size of 32 nm, purchased from Sigma Aldrich, were dissolved in ethanol to obtain colloidal suspension with concentration of 0.05 mg/ml. To improve homogeneity of the suspension, ultrasonic treatment and stirring were provided for 20 minutes. As a result, ‘milky’ suspension was prepared. Fifty microliters of the suspension were deposited on Si and glass substrates, dried and annealed in air atmosphere at 350°C for 2 hours.

2. Structural and optical characterization of TiO₂-NP-modified samples

Structural properties of TiO₂ nanostructures deposited on glass (glass/TiO₂) were studied using Raman spectroscopy (Raman spectrometer with Ar/Kr laser (Jobin Yvon- Labram 1B, $\lambda=647.1$ nm, power = 100 mW, power on the sample 9 mW). Spectral resolution of 1 cm⁻¹ was used for the registration of Raman spectra.

The surface morphology was studied by Asylum Research MFP-3D atomic force microscope, operating in tapping mode and equipped with a commercial silicon tip. The size of the AFM images was 4 μm x 4 μm .

Optical properties of glass/TiO₂ were investigated by PL. UV laser ($\lambda=266$ nm, power density 0.1 W/cm²) was used to excite PL. PL registration was performed with a monochromator (Jobin Yvon – Spex HR 460) and a CCD detector (Jobin Yvon – Spex Spectrum-1, horizontal resolution of 2000 pixels).

3. Preparation of biological samples

Autoclaved culture of *Salmonella typhimurium*, which acted as antigen (*Salmonella*-Ag), obtained from the State Scientific-Research Control Institute of Veterinary Preparations and Feed Additives (Kiev, Ukraine) was dissolved in phosphate buffer saline (PBS), pH=7.4, at concentration of 10⁹ cells per 1 ml (cl/ml). The initial solution of Ag was diluted by PBS, pH=7.4, to get concentrations of 10, 10², 10³, 10⁴, 10⁵ and 10⁶ cl/ml. ‘Anti-*Salmonella* serum’, which contained antibodies (Ab) against *Salmonella* (anti-S-Ab) was purchased from the St. Petersburg Research Institute of Vaccines and Serums (St. Petersburg, Russia) with initial concentration of 1 mg/ml. The initial solution of Ab was diluted by PBS, pH 7.4, at serum to PBS volume ratios of 1:200, 1:100, 1:50 and 1:10. Bovine serum albumine (BSA) was dissolved in PBS, pH 7.4 at concentration 1 mg/ml.

4. Fabrication and evaluation of an immunosensor

In order to determine the optimal concentration of anti-S-Ab to form glass/TiO₂/anti-S-Ab layered structure selective towards *Salmonella*, 5 μl of solution was deposited on TiO₂-modified surfaces from solutions with different concentrations of anti-S-Ab (diluted by PBS, pH 7.4, at volume ratios 1:200, 1:100, 1:50 and 1:10) and incubated at room temperature for 20 minutes. Then the anti-S-Ab-modified glass/TiO₂/anti-S-Ab surfaces were washed with PBS, pH 7.4 and gently dried with nitrogen flow.

BSA was deposited from 5 μ l of 1 mg/ml BSA solution on the surface of glass/TiO₂/anti-S-Ab to block free surface sites and to prevent glass/TiO₂/anti-S-Ab structure from non-specific interaction.

The determination of *Salmonella* was performed by step by step addition of 5 μ l of *Salmonella*-Ag solutions with concentrations 10³, 10⁴, 10⁵ and 10⁶ cl/ml. PL spectra were recorded after each modification and/or determination step.

Results and Discussion

3.1. Structural and optical properties of TiO₂ nanostructures

Fig. 1. Raman spectrum of TiO₂ nanostructures (a), AFM image of the surface of TiO₂ nanostructures (b).

Raman spectrum of TiO₂ nanostructures which were deposited on glass substrates, (glass/TiO₂) is shown in figure 1a. In this spectrum the peaks at 392, 512 and 634 cm⁻¹, which respectively correspond to B_{1g}, A_{1g} and E_g modes of anatase phase of TiO₂, are observed .

During the next experiment the surface morphology of deposited samples was investigated with AFM/SEM technique (Fig. 1b). The obtained TiO₂ nanostructure layers had high active surface area. Mean square surface roughness (RMS), measured with software Gwiddion, was 140 nm for prepared TiO₂ nanostructures.

PL spectra of TiO₂ showed wide emission band, centered at 472 nm (Fig. 2). Gaussian fitting of the PL spectrum in Origin 7.0 showed two peaks, which were centered at 461 and 502

nm. The emission at 461 nm is caused by self trapped excitons and the peak at 502 nm is observed due to the emission from oxygen vacancies .

Fig. 2. PL spectrum of glass/TiO₂ structure excited by 266 nm laser.

3.2. Effectiveness of immobilization of anti-S-Ab on the surface of glass/TiO₂ nanostructures

Effectiveness of immobilization plays an important role in the biosensor response towards the analyte. The structure of immobilized layer determines the number of active sites that are able to interact with a target analyte .

Anti-*Salmonella* antibodies (anti-S-Ab) were deposited on the surface of glass/TiO₂ from solutions containing different concentrations of anti-S-Ab. After the immobilization of anti-S-Ab, PL measurements of glass/TiO₂/anti-S-Ab structures were performed. It was found, that the immobilization of glass/TiO₂ led to an increase of the PL intensity and IR shift of the peak positions for all studied concentrations of anti-S-Ab (Fig. 3). The obtained results are summarized in figure 4, which shows the changes of both peak positions and PL intensity after incubation of glass/TiO₂ in solutions containing different concentrations of anti-S-Ab. At the lowest anti-S-Ab concentrations (1:200 and 1:100) drastic changes in PL intensity were observed. The saturation of PL intensity, caused by anti-S-Ab immobilization, was observed after incubation in the samples with the highest concentrations of anti-S-Ab (using 1:50 and 1:10 diluted initial sample of anti-S-Ab). The most significant PL intensity changes vs concentration of anti-S-Ab were observed after the incubation of glass/TiO₂ in 1:100 diluted initial sample of anti-S-Ab.

The shift of the PL peak position increased from 4 to 8 nm after incubation of glass/TiO₂ in lower concentrations of anti-S-Ab and this effect become saturated after incubation in higher concentrations of anti-S-Ab.

Incubation of glass/TiO₂/anti-S-Ab structures in 1 mg/ml solution of BSA resulted in the increase of the PL intensity for all studied glass/TiO₂/anti-S-Ab/BSA samples (Fig. 3).

Fig. 3. PL spectra of glass/TiO₂ structure before and after the immobilization of anti-S-Ab and incubation in 1 mg/ml BSA solution: a) glass/TiO₂ structure incubated in 1:200 diluted initial sample of anti-S-Ab, b) glass/TiO₂ structure incubated in 1:10 diluted initial sample of anti-S-Ab.

Fig. 4. The changes of glass/TiO₂ spectra after the immobilization of anti-S-Ab from solutions containing different concentrations of anti-S-Ab.

3.2. Sensitivity of glass/TiO₂/anti-S-Ab/BSA structures to *Salmonella-Ag*

Glass/TiO₂/anti-S-Ab/BSA structures were incubated in *Salmonella*-Ag solutions with concentrations of 10³ – 10⁶ cells/ml (cl/ml) (Fig. 65). A decrease of the PL intensity of glass/TiO₂/anti-S-Ab/BSA structures was observed after the incubation in *Salmonella*-Ag sample due to the formation of immune complex between antigens and antibodies. It was found that the PL spectra did not change after the incubation of glass/TiO₂/anti-S-Ab/BSA structures in *Salmonella*-Ag solutions of 10⁴ – 10⁵ cl/ml for structures with analyte-sensitive layers (anti-S-Ab/BSA), formed from highest applied concentrations of anti-S-Ab. The hyperbolic function of the PL signal vs concentration of analyte is related to the ‘saturation’ of the bio-recognition layer by analyte at their higher concentrations .

Fig. 5. PL spectra during the determination of *Salmonella*-Ag by glass/TiO₂/anti-S-Ab/BSA based immunosensor.

Procedures applied for the determination of *Salmonella* antigens by glass/TiO₂/anti-S-Ab/BSA-based immunosensor consist of incubation in *Salmonella* antigens containing solution (20 min),

washing out of unbounded interfering materials (3-6 min.) and the PL-measurement (1 min.). These procedures all together last less than 30 min.

3.3. Mechanism of PL signal evolution of glass/TiO₂/anti-S-Ab/BSA structure response to *Salmonella*-Ag

Results of sensitivity tests to *Salmonella*-Ag were analyzed for TiO₂ samples with glass/TiO₂/anti-S-Ab/BSA-based biosensitive layers, formed from 1:200 dissolved initial samples of anti-S-Ab (Fig. 6a). The sensitivity was calculated according to the formula:

$$S = \frac{I_0 - I_1}{I_0} \quad (1)$$

where I_0 and I_1 are PL intensities of glass/TiO₂/anti-S-Ab/BSA structure at peak positions registered before (I_0) and after (I_1) incubation in *Salmonella*-Ag containing solution. The sensitivity of TiO₂ biosensor has a linear slope in semi-logarithmic scale in the range of 10³ – 10⁵ cl/ml.

Fig. 6. Sensitivity of glass/TiO₂/anti-S-Ab/BSA structure formed using 1:200 diluted anti-S-Ab towards *Salmonella*-Ag (a), ratio of $I_{STE}/I_{V[O]}$ after each step of bio-functionalization and sensitivity tests after incubation of TiO₂/anti-S-Ab/BSA in solution containing *Salmonella* antigens at concentrations of 10³, 10⁴, 10⁵, 10⁶ cells/ml, respectively: TiO₂/anti-S-Ab/BSA/Ag-10³, TiO₂/anti-S-Ab/BSA/Ag-10⁴, TiO₂/anti-S-Ab/BSA/Ag-10⁵, TiO₂/anti-S-Ab/BSA/Ag-10⁶ (b), average values of PL intensity at PL maximum ($\lambda=471$ nm) of TiO₂ nanostructures after each step of bio-functionalization and sensitivity tests after incubation of TiO₂/anti-S-Ab/BSA in solution containing *Salmonella* antigens at concentrations of 10³, 10⁴, 10⁵, 10⁶ cells/ml (c), PL signals of glass/TiO₂/anti-S-Ab and glass/TiO₂/anti-S-Ab/BSA structures after the incubation in *Salmonella*-Ag sample containing 10⁶ cl/ml (d).

As it was shown before, the PL spectrum of glass/TiO₂ structure can be divided into two spectra with corresponding peaks. These spectra are related to PL induced by recombination of: (i) self trapped excitons (STE) and (ii) oxygen vacancies. We performed the fitting of PL spectra of glass/TiO₂ structure before and after interaction with anti-S-Ab, BSA and *Salmonella*-Ag (the fitting is not shown here) and plotted the ratio on PL intensities related to STE and oxygen vacancies (Fig. 6b).

After the immobilization of *Salmonella*-Ab, which was followed by BSA adsorption, the PL intensity of glass/TiO₂/anti-S-Ab/BSA induced by oxygen vacancies has increased in comparison to that of glass/TiO₂/anti-S-Ab, what could be chosen (or induced) by several reasons: (i) the decrease of recombination of excitons (ii) charge transfer or (iii) electrostatic interaction. The formation of anti-S-Ab/*Salmonella*-Ab-based immune-complex was assisted by the decrease of PL mode, which was related to oxygen vacancies and the ratio of $I_{STE}/I_{V[O]}$ reached the saturation at high concentrations of *Salmonella*-Ag.

The stability (repeatability) of the obtained immunosensor has been evaluated and shown in figure 6c that represents the statistics data of all immunosensor tests. The tests were performed for 7 TiO₂ samples. The average values of PL intensity after each step of bio-functionalization of TiO₂ nanoparticles and sensitivity tests of glass/TiO₂/anti-S-Ab/BSA structure to *Salmonella* antigens have been calculated as the standard deviation between number of different measurements. The studied samples showed almost similar sensitivity towards anti-*Salmonella* antibodies, BSA and target *Salmonella* antigens molecules. Some deviations were observed due to the insignificant variations in the surface area of TiO₂, which may influence a surface-density of immobilized proteins and/or some other variations of here applied analytical system.

Supplementary PL-based test was performed in order to evaluate the specificity of the glass/TiO₂/anti-S-Ab structure (Fig. 6d). It was found that after BSA deposition no drastic differences were observed in the PL spectra of glass/TiO₂/BSA and glass/TiO₂/ BSA/*Salmonella*-Ag structures. Significant changes in PL signals of glass/TiO₂/anti-S-Ab and glass/TiO₂/anti-S-Ab/BSA structures before and after incubation in solutions containing *Salmonella*-Ag can point to the specificity of immune interaction of antigens (*Salmonella*-Ag) with antibodies (anti-S-Ab), which are present in glass/TiO₂/anti-S-Ab and glass/TiO₂/anti-S-Ab/BSA structures.

Possible mechanisms of interaction between TiO₂ and biological compounds, which were included in anti-S-Ab, BSA and *Salmonella*-Ag aliquots, could be explained by passivation of surface states, charge-transfer, electrostatic interaction and Förster energy transfer . As the anti-S-Ab, BSA and *Salmonella*-Ag are optically transparent in the UV/Vis range of the wavelengths their optical density can be neglected . Moreover, for these compounds the charge-transfer cannot be considered because neither anti-S-Ab nor BSA are able to attend in electron transfer reactions.

The most obvious mechanism of anti-S-Ab adsorption on TiO₂ is based on electrostatic interaction between the surface charge of TiO₂ and the charge of anti-S-Ab proteins and van der Waals interaction. Anti-S-Ab antibodies, as all the proteins, consist of amino acids that might contain positively and/or negatively charged radicals that are determining the charge of different protein domains . The large amount of negatively charged groups such as carboxyl (-COOH), aldehyde (-CHO), hydroxyl (-OH), positively charged primary amine (-NH₂), and some other groups, which are involved into the structure of amino acids, are responsible for the partial charges (δ^+ and δ^-) of some protein domains. Naturally, the charges at least partly are compensating each other, but since the ternary structure of proteins is relatively rigid and the charged groups have only limited degree of freedom to move within the protein, in some parts of the protein some uncompensated charge on the surface and inside the protein still remains . Due to the latter effect, anti-S-Ab proteins contain some partially charged groups and domains with partial positive charge ' δ^+ ' and negative charge ' δ^- ', which are more or less randomly located on the surface and inside of the protein [40], the localization of these charged groups/domains depends on the amino acid sequence [39, 40]. The partial charge of each such group is mostly lower than the full charge of an electron e^- (1.6×10^{-19}), but anyway it allows relatively strong electrostatic interaction with oppositely charged surface domains of TiO₂. Such electrostatic effects can be considered as the main reason of the interaction between TiO₂ and anti-S-Ab proteins in addition to the attraction of van der Waals forces, hydrogen and disulfide bonds, etc. The TiO₂ (anatase) is known as n-type semiconductor with an 'upward bending' of LUMO-, HOMO-related and even some other electron energetic levels [41] suggesting the negatively

charged surface, consequently, the positive partial uncompensated charge of anti-S-Ab (δ^+) interacts electrostatically with the surface charge of TiO₂. Further interaction of glass/TiO₂/anti-S-Ab/BSA-based biosensitive layer with target analyte (*Salmonella*-Ag) leads to both the UV-shift of PL peak positions and the decrease of PL intensity to the value that corresponds to the pristine TiO₂. The latter effect is caused by the fact that during the formation of an immune complex between immobilized anti-S-Ab and *Salmonella*-Ag, which are present in tested aliquot, the uncompensated charges (δ^- and δ^+) of both proteins are involved in the electrostatic interactions between oppositely charged groups in the structures of both: *Salmonella*-Ag and anti-S-Ab. As a result, the part of the charged groups, which initially were involved in the interaction between the anti-S-Ab and TiO₂, are compensated by the oppositely charged domains/groups of the *Salmonella*-Ag proteins, in such a way reducing the electrostatic influence of anti-S-Ab to PL centers on the surface of TiO₂. Due to these effects significant changes in the positions and intensity of PL peaks are observed before and after incubation of glass/TiO₂ – modified surfaces in solutions containing here evaluated biological compounds (anti-S-Ab, BSA and *Salmonella*-Ag).

Conclusions

The photoluminescence of TiO₂ nanoparticles deposited on glass substrate (glass/TiO₂) were applied for the detection of *Salmonella*-Ag. Glass/TiO₂/anti-S-Ab sensitive layer was formed and applied for the detection of *Salmonella*-Ag. More selective biological recognition layer was formed by treatment of glass/TiO₂/anti-S-Ab by BSA. The optimal concentration of anti-S-Ab in a solution used for the formation of glass/TiO₂/anti-S-Ab structure was in 1:100 diluted initial sample of anti-S-Ab. The immobilization of anti-S-Ab on the TiO₂ surface resulted in the increase of PL intensity and significant IR-shift of the PL spectra of glass/TiO₂/anti-S-Ab structure in comparison with that of glass/TiO₂ structure. The interaction of glass/TiO₂/anti-S-Ab/BSA biological recognition layer with *Salmonella*-Ag resulted in the decrease of PL intensity and UV-shift of PL spectra. The sensitivity of the obtained optical biosensor is in the range from 10³ – 10⁵ cl/ml of *Salmonella*-Ag. The specificity and sensitivity of glass/TiO₂/anti-S-Ab/BSA-based immunosensor has been additionally proved by evaluation of PL of glass/TiO₂/BSA and glass/TiO₂//BSA/*Salmonella*-Ag structures. It was found that after the deposition of BSA on the surface of glass/TiO₂ (formation of glass/TiO₂/BSA structure) followed by similar deposition of *Salmonella*-Ag (formation of glass/TiO₂/BSA/*Salmonella*-Ag structure) no drastic differences

were observed in the PL spectra of glass/TiO₂/BSA and glass/TiO₂/ BSA/*Salmonella*-Ag structures.

Obviously, the interaction between TiO₂ and biological compounds (anti-S-Ab, BSA and *Salmonella*-Ag) is based on electrostatic interactions, in addition to the van der Waals forces, sulfide bonds and hydrogen bonds, which all take place during the protein adsorption on TiO₂ and the formation of anti-S-Ab/*Salmonella*-Ag immune complex.

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